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(54) Rubber composition

(57) A rubber composition containing, per 100 parts by weight of a vulcanizable rubber component, 0.2 to 20 parts by weight of an ester compound obtained by a reaction of alcohols with a fatty acid having a conjugated linoleic acid content of 40 % by weight or more, which is markedly excellent in processability in the unvulcanized state and which has an excellent low heat penerating property without deteriorating mechanical properties after vulcanization and is excellent in adhesion between steel and vulcanized rubber.

BACKGROUND OF THE INVENTION

5 1. Field of the Invention

[0001] The present invention relates to a rubber composition which contains a fatly acid ester and is improved in processability without deteriorating mechanical properties after vulcanization. Further, the present invention relates to a rubber composition which is excellent in adhesion between vulcanized rubber and steel and the like without deteriorating mechanical properties after vulcanization and which has good processability.

2. Description of the Related Art

[0002] In recent years, a rubber composition for a tire is notably deteriorated in processability of unvulcanized rubtis ber in compensation for satisfying desirable performances as a tire over a wide range.

[0003] In general, it is known that a process oil is blended into rubber for the purpose of providing diene base rubber with plasticity to improve processability.

[0004] However, involved in this method is a problem that while unvulcanized rubber can be improved in processability, vulcanized rubber is reduced in mechanical properties and the like.

20 [0005] On the other hand, known are rubber compositions containing fatty acids and various metal salts of fatty acids as processing aids (Japanese Patent Application Laid-Open No. Sho 65-113044, etc.). They can be expected to improve the processability but have problems in that the mechanical properties after vulcanization are reduced in terms of the use of fires and that durability in adhesion between a steel cord and vulcanized rubber is deteriorated so that they can not be used for the circumference of a steel cord and are restricted in use.

[0006] Further, used as processing aids are various esters, for example, pentaerythritol tetrastearate (Japanese Patent Application Laid-Open No. Hel 1-115943), monoglycerides, rice bran wax and paraffilm wax. While they can be expected to improve the processability, they have a problem of blooming or bloeding in vulcanized rubber.

[0007] Esters of a polymer acid or an alkenylsuccinic acid with a higher alcohol or a polyhydric alcohol are disclosed in Japanese Patent Application Laid-Open No. Sho 62-72733 as an ester base processing aid which solves these problems, but they reduce the mechanical properties after vulcanization and have a problem of durability in adhesion between steel and vulcanized rubber.

[0008] Further, a processing aid which is a mixture of a diester of alighatic alcohol having a melting point of 10°C lower with an organic acid, and a fatty acid is disclosed in Japanese Patent Application Laid-Open No. Sho 57-58 168998, but this reduces as well the mechanical properties after vulcanization and has a problem of duability in adhesion between steel and vulcanized nubber. Further, found are no descriptions on the effect of adding respective components contained in the processing aids in the form of a simple substance to a rubber composition.

[0009] Further, esters of an unsaturated fatty add with a saturated or unsaturated alcohol or glycol are disclosed in Japanese Patent Application Laid-Open No. Sho 59-122532 as an agent to improve resistance to low temperature of a nubber composition. This agent can improve flexibility of the rubber composition but the rubber composition containing the agent has a problem that it is inferior in a low heat-generating property, strength and an adhesion property to a steel cond.

[0010] As described above, the existing situation is that respective conventional rubber compositions which are improved in processability have the problems that the physical properties such as mechanical properties and a low heat-generating property are largely reduced and the adhesion between steel and vulcanized rubber is deteriorated, and a rubber composition which is improved in these problems and has good processability is desired.

SUMMARY OF THE INVENTION

50 [0011] In light of the problems on the conventional techniques described above and the existing situation, the present invention is intended solving them, and an object thereof is to provide a rubber composition which is excellent in processability in the unvulcanized state and has an excellent low heat-generating property without deteriorating mechanical properties after vulcanization and which is excellent in adhesion between steel and vulcanized rubber.

[0012] Intensive investigations of the problems described above continued by the present inventors have resulted in finding that a rubber composition which is blended with a specific ester compound is notably excellent in processes billty in the unrulcanized state and has an excellent low heat-generating property without deteriorating mechanical properties after vulcanization and that it is excellent as well in adhesion between steel and vulcanized nubber, and thus the present invention has been completed.

[0013] That is, the rubber composition of the present invention which has good processability comprises the following items (1) to (3):

- (1) A rubber composition characterized by comprising, per 100 parts by weight of a vulcanizable rubber component, 0.2 to 20 parts by weight of an ester compound obtained by a reaction of an alcohol with a fatty acid having a conjugated finoleic acid content of 40 % by weight or more.
- (2) The rubber composition as described in the above item (1), wherein the alcohol contains 60 % by weight or more of monohydric alcohol and/or dihydric alcohol.
- (3) The rubber composition as described in the above item (1) or (2), wherein the ester compound has a viscosity of 140 mPa s or less at 25°C.

DETAILED DSCRIPTION OF THE PREFERRED EMBODIMENTS

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[0014] The embodiments of the present invention shall be explained below in detail.

[0015] The rubber composition of the present invention is characterized by containing, per 100 parts by weight of a vulcarizable rubber component. Oz to 20 parts by weight of an ester compound obtained by a reaction of an alcohol with a fatty acid having a conjugated lincelize acid content of 40% by weight for more.

[0016] The nubber component used in the present invention shall not specifically be restricted as long as it is vulcanizable rubber and, for example, natural rubber (NR), styrene-butdiene rubber (SBR), polybutadiene rubber (BR) and isoprene rubber (IR) can be used alone or in combination of two or more kinds thereof.

[0017] The ester compound used in the present invention is obtained by esterification of a fatty acid having a conjugated linoleic acid content of 40 % by weight or more with an alcohol.

[0018] The esterification can be carried out by a dehydration reaction of a fatty acid with an alcohol.

[0019] The fatty acid having a conjugated linoleic acid content of 40 % by weight or more, which is the raw material, can be obtained primarily by the following processes (1) to (3):

- (1) A process for producing dehydrated castor oil fatty acid by a dehydration reaction of castor oil fatty acid.
- (2) Modification to dehydrated castor oil fatty acid by a saponification decomposition of dehydrated castor oil.
- (3) Conjugation isomerization of a fat and oil having a high linoleic acid content such as safflower oil or the fatty acid thereof.

[0020] The fatty acid having a conjugated linoleic acid content of 40 % by weight or more can be obtained by the processes (1) to (3) described above, but the process for obtaining the fatty acid shall not be restricted to these processes.

[0021] In the present invention, any fatty acids can be used without specific restrictions as long as they are fatty acids having a conjugated linoleic acid content of 40 % by weight or more, but fatty acids having a conjugated linoleic acid content of 45 % by weight or more are viewpoint of elevating reactivity with a polymer via sulfur.

[0022] The fatty acids having a conjugated linoleic acid content of less than 40 % by weight reduce the reactivity described above, which in turn leads to bleeding and a reduction in the physical properties of vulcanized rubber, and therefore the var en ot preferred.

[0023] The alcohols which can be used in the present invention include monohydric alcohols, dihydric alcohols and polyhydric alcohols of trihydric or higher alcohols. In order to obtain the ester compounds having a low viscosity, monohydric alcohols or dihydric alcohols or mixtures thereof are suited.

[0024] Particularly preferably, monohydric alcohols or dihydric alcohols or mixtures thereof account for 60 % by weight or more of the alcohols used.

[0025] Polyhydric alcohols of trihydric or higher alcohols can be used in combination within a range where the resulting ester compound has a viscosity which does not exceed 140mPa • s at 25°C.

[0026] There can be used as the monohydric alcohol, for example, methyl alcohol, ethyl alcohol, isopropyl alcohol, butyl alcohol, isotyl alcohol, isotyl alcohol, isotyl alcohol, isotyl alcohol, isotyl alcohol, benzyl alcohol, see 2 -ethylheyyl alcohol, disotyl alcohol, isotyl alcohol, alcohol, alcohol, alcohol, see 3 -ethylheyyl alcohol, disotyl alcohol, alcoho

[0027] There can be used as the dihydric alcohol, for example, ethylene glycol, propylene glycol, butylene glycol, hex/lene glycol, diethylene glycol, neopentyl glycol and N-methyldiethanolamine.

hexylene glycol, diethylene glycol, neopentyl glycol and N-methyldiethanolamine.
[0028] For example, trimethyloipropane can be used as the trihydric alcohol.

[0029] The viscosity of the ester compound comprising the fathy and having a conjugated linoleic add content of 40 % by weight or more and an alcohol in the present invention varys depending on the kinds of the fathy acid and acohol in the present invention are varys depending on the kinds of the fathy acid and acohol used, and it should be low [5 to 200 mPa * s at 25°C, more preferably 5 to 100 mPa * s at 25°C from a viewpoint of a reduction in the viscosity of unvulcanized rubber after blended with the ester compound.

[0030] The blending amount of the ester compound is 0.2 to 20 parts by weight, preferably 0.5 to 5 parts by weight per 100 parts by weight of the vulcanizable rubber component described above.

[0031] If the blending amount is less than 0.2 part by weight, improvement in the processability is less likely to be expected. On the other hand, if it exceeds 20 parts by weight, the physical properties of the rubber tend to be reduced, and it is not preferred from a viewpoint of the cost.

[0032] Further, the rubber composition of the present invention is preferably compounded with a reinforcing agent. The reinforcing agent shall not specifically be restricted, and those which are generally used for rubber can be applied. Included are, for example, carbon black such as HAF, ISAF and SAF, or white fillers such as silica and aluminum hydroxide.

[0033] Further, the rubber composition of the present invention can suitably be blended, if necessary, with other rubber compounding ingredients which are generally used in the rubber industry, for example, suffur, a vulcanization accelerator, a process oil and an antioxidation.

(1034) The blending amounts of the reinforcing agents and the rubber compounding ingredients described above shall suitably be determined respectively within the ranges where the effects of the present invention are not damaged.

[1035] The ester compound in the present invention can reduce the viscosity of the rubber compounds to the present invention can reduce the viscosity of the rubber compound in the present invention can reduce the viscosity of the rubber compounding ingredients and the like). Further, it has a hydrophilic group (ester group) and a lipophilic group (fath said) in a molecule, so that the hydrophilic group adsorbs on the surface of the reinforcing agent and the lipidic group is compatible with rubber, whereby interaction between the reinforcing agent and the process and the rubber in the unvalcanized state is 20 reduced. As a result, the rubber composition is improved in processability and the reinforcing agent can be dispersed well.

[0036] Further, the ester compound of the present invention can react with the polymer via sulfur during vulcanization, so that the mechanical properties of vulcanized rubber are not deteriorated, and blooming or bleeding are less liable to take place.

[0037] The rubber composition of the present invention is obtained by kneading the rubber component, the ester compound, the reinforcing agent and the rubber compounding ingredients each described above by means of a mixer such as a roll, a kneader and a barbury mixer, then mold-processed, and vulcanized. It can be used for the use such as a tire tread, a carcass, a belt, a side wall and a bead part, rubber for a conveyor belt, and other various industrial rubbers.

[0038] In particular, the rubber composition of the present invention is, as described above, markedly excellent in processability in the unvulcanized state and has an excellent low heat-generating property without deteriorating the mechanical properties after vulcanization. Further, it is excellent as well in adhesion between steel and vulcanized rubber, and therefore it is suitably used for the use such as a carcass, a belt, a bead part and the like.

[0039] Further, in order to improve the adhesion to steel, it is effectively blended with an adhesive such as a cobalt salt of an organic acid or an inorganic salt hydrate in an amount of, for example, 0.7 to 1.0 part by weight.

EXAMPLES

[0040] The present invention shall be explained below in further details with reference to examples and comparative examples, but the present invention shall not be restricted to the following examples.

Examples 1 to 9 and Comparative Examples 1 to 6

[0041] Rubber compositions were prepared in the respective blending formations using ester compounds A to K obtained by reacting fatty acids having various conjugated linoleic acid contents with various alcohols shown in the following Table 1, and natural rubber and polybutadiene rubber as vulcanizable rubber components.

Examples 10 to 18 and Comparative Examples 7 to 12

[0042] Rubber compositions were prepared in the respective blending formations shown in the following Table 2 using ester compounds A to K obtained by reacting fattly acids having various conjugated linoleic acid contents with various alcohols, and natural rubber as a vulcanizable rubber component.

[0043] The conjugated linoleic acid contents of the various fatty acids used for ester compounds A to K described above and the viscosities of the esters were measured by a UV spectrum method and a Brookfield type rotary viscom-5 eter, respectively.

[0044] Ester compounds A to K were obtained by production methods described below. Hereinafter, "conjugated linoleic acid content" is referred to simply as "conjugated acid rate".

A: A one litter flask was charged with 364 g (1.3 mol) of dehydrated castor oil fatty acid (conjugated acid rate: 45 %), 62 g (0.69 mol) of 1,4-butylene glycol, 100 ml of toluene and 0.4 g of arrhydrous stannous chloride as a catalyst, and reaction was cortinued at 140 to 160°C for 8 hours while removing generated water to the outside of the system. After cooling the reaction liquid to 120°C, 5 g of clay was added thereto to treat it for 10 minutes, and then it was filtered to obtain an ester compound having an acid value of 2.3 and a viscosity at 25°C of 54 mPa *s. This ester compound had a conjugated acid rate of 43 %.

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B: A one liter flask was charged with 280 g (1.0 mol) of dehydrated castor oil fatty acid (conjugated acid rate: 45 %), 169 g (1.3 mol) of 2-ethylhexanol, 100 ml of toluene and 0.3 g of paratoluenesulfonic acid, and reaction was continued at 140 to 160°C for 8 hours while removing generated water to the outside of the system. After distilling toluene and excess unreacted alcohol to recover them, the residue was treated with acid day to obtain an ester compound having an acid value of 0.7 and a viscosity at 25°C of 14 mPa * s. This ester compound had a conjugated acid rate of 43 %.

C: A one liter flask was charged with 280 g (1.0 mol) of dehydrated castor oil fatty add (conjugated acid rate: 45 %), 130 g (1.2 mol) of benzyl alcohol, 100 ml of toluene and 0.3 g of paratoluenesulfonic acid, and reaction was continued at 140 to 160°C for 8 hours while removing generated water to the outside of the system. After distilling toluene and excess unreacted alcohol to recover them, the residue was treated with acid clay to obtain an ester compound having an acid value of 0.8 and a viscosity at 25°C of 16 mPa * s. This ester compound had a conjugated acid rate of 44 %.

D: A one liter flask was charged with 280 g (1.0 mol) of dehydrated castor oil fatty add (conjugated acid rate: 45 %), 192 g (6.0 mol) of methyl alcohol and 0.3 g of paratoluenesutloric acid, and reaction was continued for 16 hours under reflux. After distilling the excess urreacted alcohol to recover it, the residue was distilled under reduced pressure to obtain an ester compound having an acid value of 0.3 and a viscosity at 25°C of 7 mPa •s. This ester compound had a conjuacted acid rate of 41 %.

E: A one liter flask was charged with 384 g (1.3 mol) of dehydrated castor oil fatty add (conjugated acid rate: 45 %), 83 g (0.78 mol) of diethylene glycol, 100 ml of toluene and 0.4 g of anhydrous stannous chloride as a catalyst, and reaction was continued at 140 to 160°C for 12 hours while removing generated water to the outside of the system. After cooling the reaction liquid to 120°C. 5 g of clay was added thereto to treat it for 10 minutes, and then it was filtered to obtain an ester compound having an acid value of 6.9 and a viscosity at 25°C of 58 mPa * s. This seter compound had a conjugated acid rate of 42 %.

F: A one liter flask was charged with 364 g (1.3 mol) of dehydrated castor oil fatty acid (conjugated acid rate: 45 %), 59 g (0.44 mol) of trimethylolpropane, 100 ml of toluene and 0.4 g of anhydrous stannous chloride as a catalyst, and reaction was confinued at 150 to 180°C for 5 hours and then at 180 to 195°C for 7 hours while removing generated water to the outside of the system. After cooling the reaction liquid to 120°C, 5 g of clay was added thereto to freat it for 10 minutes, and then it was filtered to obtain an esteri compound (frester) having an acid value of 6.8 and a viscosity at 25°C of 160 mPa • s. This triester compound had a conjugated acid rate of 45 %.

A mixture obtained by mixing this triester compound with ester compound A in a proportion of 30: 70 was used.

G: A one liter flask was charged with 364 g (1.3 mol) of dehydrated castor oil fatty acid (conjugated acid rate: 50 %), 62 g (0.69 mol) of 1,4-butylene glycol, 100 mil of toluene and 0.4 g of arhydrous stannous critoride as a catalyst, and reaction was continued at 140 to 160°C for 8 hours while removing generated water to the outside of the system. After cooling the reaction liquid to 120°C, 5 g of clay was added thereto to treat it for 10 minutes, and then it was filtered to obtain an ester compound having an acid value of 2.3 and a viscosity at 25°C of 54 mPa * s. This ester compound had a conjugated acid rate of 48 %.

H: A one liter flask was charged with 367 g (1.3 mol) of oleic acid, 82 g (0.69 mol) of 1.4-butylene glycol, 100 ml of toluene and 0.4 g of anhydrous stannous chloride as a catalyst, and reaction was continued at 140 to 160°C for 8 hours while removing generated water to the outside of the system. After cooling the reaction liquid to 120°C, 5 g of clay was added thereto to treat it for 10 minutes, and then it was filtered to obtain an ester compound having an acid value of 2.4 and a viscosity at 25°C of 55 mPa • s.

I: A one liter flask was charged with 282 g (1.0 mol) of oleic acid, 169 g (1.3 mol) of 2-ethylhexanol, 100 ml of toluene and 0.3 g of paratoluenesulfonic acid, and reaction was continued at 140 to 160°C for 8 hours while removing

generated water to the outside of the system. After distilling toluene and the excess unreacted alcohol to recover them, the residue was treated with acid clay to obtain an ester compound having an acid value of 0.8 and a viscosity at 25°C of 16 mPa *s.

J: A one liter flask was charged with 282 g (1.0 mol) of oleic acid, 130 g (1.2 mol) of berzyl alcohol, 100 ml of toluene and 0.3 g of parabiouenesultonic acid, and reaction was continued at 140 to 160°C for 8 hours white removing generated water to the outside of the system. After distilling toluene and the excess unreacted alcohol to recover them, the residue was treated with acid clay to obtain an ester compound having an acid value of 0.9 and a viscosity at 25°C of 18 mPa • s.

K: A one liter flask was charged with 364 g (1.3 mol) of dehydrated castor oil fatty acid (conjugated acid rate: 35%), 62 g (0.69 mol) of 1,4-butylene glycol, 100 ml of toluene and 0.4 g of arhydrous stannous chloride as a catalystand reaction was continued at 140 to 160° for 8 hours while removing generated water to the outside of the system. After cooling the reaction liquid to 120°C, 5 g of clay was added thereto to treat it for 10 minutes, and then it was filtered to obtain an ester compound having an acid value of 2.4 and a viscosity at 25°C of 55 mPa • s. This ester compound had a conjugated acid rate of 33 %.

[0045] The rubber compositions obtained in Examples 1 to 18 and comparative Examples 1 to 12 were evaluated for physical properties of the unvulcanized state in terms of a Mooney viscosity, and the vulcanized state in terms of a heat-generating property (resilience) and tensile strength (T_B). The adhesive property between a steel cord and vulcanized rubber was evaluated by an index which was derived from a peel-off strength obtained by a steel cord • peel off test. These results are shown in the following Table 1 and Table 2.

[0046] The Mooney viscosity, the heat-generating property (resilience), the tensile strength (T_B) and the steel cord-peel off test were evaluated by the following standards, respectively. Vulcanization was carried out at 145°C for 30 minutes.

(1) Evaluation method of Mooney viscosity

[0047] The Mooney viscosity (ML₁₋₄, 130°C) was determined according to JIS K6300-1994. It is shown that the lower the value, the better the processability in the unvulcanized state.

(2) Evaluation method of heat-generating property (resilience)

[0048] The resilience was determined at room temperature (25°C) according to the rebound resilience test prescribed in JIS K6301-1995. It is shown that the higher the value is, the better the low heat-generating property is.

(3) Evaluation method of tensile strength (T_B)

[0049] The tensile strength was determined at room temperature (25°C) according to the tensile test prescribed in 40 JIS K6301-1995. It is shown that the higher the value is, the better the tensile strength is.

(4) Evaluation method of steel cord · peel off test (adhesive property)

[0050] With regard to an adhesive property between a steel cord and vulcanized rubber, a peel-off strength (peeling strength) between a steel cord and vulcanized rubber was determined by peeling off the steel cord after embedding the steel cord in the unvulcanized rubber composition to vulcanize at 145°C for 45 minutes, and then aging it in a constant temperature bath of 100°C for 7 days.

[0051] The values were shown by indexes wherein the value of Comparative Example 1 or Comparative Example 7 was set at 100. It is shown that the higher the value is, the better the adhesive property is.

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Table 1

				(p)	end ur	(blend unit: parts by weight)	irts b	y weig	ht)	
	Comparative					Example	6			
	Example 1	7	7	3	4	ហ	9	7	8	6
Natural rubber	09	09	09	09	09	09	09	9	9	09
Polybutadiene rubber	40	40	40	40	40	40	40	40	40	40
Carbon black (HAF)	75	75	75	75	7.5	75	7.5	75	75	75
Process oil	. 2	2	2	7	7	2	7	2	2	2
Stearic acid	1	1	1	1	1	1	1	п	1	-
Processing Kind	None	K	8	Э	q	2	A	Ð	Ą	A
aid Added amount	0	2	2	2	2	2	2	2	0.5	5
Zinc oxide	20	3	5	r	5	ß	ĸ	2	5	5
Vulcanization	-		•	•		•		•	•	•
accelerator (TBBS)	1	•	•]	•	•	•	•	,	•	•
Insoluble sulfur	4	4	4	4	4	4	4	4	4	4
Processability (ML1.4)	85	72	72	72	70	73	76	72	77	67
Heat-generating	a u	5 8 2	200	4 8 4	0 74	0	4 82	58.2	4	1.85
property (resillence)	0.00		?	:	;	2				
Strength (TB, MPa)	17.0	17.2	17.3	17.1	16.9	17.3 17.1 16.9 17.4 17.0 17.6 17.2	17.0	17.6	17.2	17.2

ß - 83

Table 1 (continued)

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parts by weight) unit: (blend

58.0 : 9 40 Example S 9 40 2 16. 2 × N S 57 57.4 16.2 Comparative 9 40 75 75 10 -N 2 0 57.6 ۰. m 9 н 4 16. 57.4 0 9 40 75 15. H 0 S н 2 amount (resilience) Processability (ML144) rubber (HAF) MPa) accelerator (TBBS) Added Kind Insoluble sulfur Heat-generating rubber Polybutadiene Strength (TB, Vulcanization Carbon black Stearic acid Process oil Processing Zinc oxide property Natural ald

The kinds of the processing acids in the above table are shown below: [0052]

- An ester compound of dehydrated castor oil fatty acid having a conjugated linoleic acid content of 45 % with A: 1.4-butviene givcol
- An ester compound of dehydrated castor oil fatty acid having a conjugated linoleic acid content of 45 % with B: 2-ethylhexanol
- 45 An ester compound of dehydrated castor oil fatty acid having a conjugated linoleic acid content of 45 % with C:
 - benzyl alcohol An ester compound of dehydrated castor oil fatty acid having a conjugated linoleic acid content of 45 % with D:
 - methyl alcohol An ester compound of dehydrated castor oil fatty acid having a conjugated linoleic acid content of 45 % with
- E: diethylene alvcol
- A mixture of ester compounds of dehydrated castor oil fatty acid having a conjugated linoleic acid content of F: 45 % with trimethyloloropane or 1,4-butylene glycol (30 % by weight/70 % by weight)
- An ester compound of dehydrated castor oil fatty acid having a conjugated linoleic acid content of 50 % with G: 1,4-butylene glycol
- An ester compound of oleic acid with 1,4-butylene glycol H
 - An ester compound of oleic acid with 2-ethylhexanol 1:
 - An ester compound of oleic acid with benzyl alcohol J:

K: An ester compound of dehydrated castor oil fatty acid having a conjugated linoleic acid content of 35 % with 1,4-butylene glycol

TBBS: N-t-butyl-2-benzothiazolyl sulfenamide

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				回	end ur	it: p	(blend unit: parts by weight	/ weig	ht)	
	Comparative					Example	9			
	Example 7	10	11	12	13	14	15	16	17	18
Natural rubber	100	100	100	100	100	100	100	100	100	100
Carbon black (HAF)	50	50	20	50	95	09	50	20	20	20
Adhesive	0.8	0.8	0.8	0.8	8.0	8.0	0.8	0.8	9.0	9.0
Antioxidant (6PPD)	2	2	2	2	2	2	2	2	2	2
Processing Kind	None	Ą	В	Ü	D	9	ы	ტ	Ą	¥
ald Added amount	0	7	2	2	2	2	2	7	0.5	2
Zinc oxide	8	8	. 8	8	8	8	8	8	8	8
Vulcanization accelerator (DCBS)	1	1	1	1	1	1	1	1	1	1
Insoluble sulfur	ις.	2	2	e,	2	5	2	5	5	2
Processability (ML1+4)	70	57	57	57	55	58	61	57	62	53
Heat-generating property (resilience)	66.2	66.2	66.3 66.0	0.99	62.9	66.0	66.0 66.0	66.4	66.0	66.1
Strength (TB, Mpa)	19.8	19.9	20.0	19.6	19.7	19.6	20.0 19.6 19.7 19.6 20.0 20.1	20.1	20.0	20.3
Adhesive property (index)	100	66	101	101	102	100	102	106	100	100

Table 2 (continued)

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(blend unit: parts by weight)

intun nuara)	1111	Darie Dy	Y WELGILL	17.1	- 1
	υ	ompare	tive I	Comparative Example	_ n l
	8	6	10	11	
Natural rubber	100	100	100	100	
Carbon black (HAF)	20	20	50	50	
Adhesive	8.0	9.0	9.0	0.8	
Antioxidant (6PPD)	2	2.	2	2	
Processing Kind	H	1	J	K	
aid Added amount	2	2	2	7	- 1
Zinc oxide	8	8	8	8	,
Vulcanization	٦	г	1	1	
Insoluble sulfur	5	5	5	ıc	1
Processability (ML,,,)	09	60	9	67	- 1
Heat-generating	63.9	63.5	64.1	65.5	-
Strength (TB, Mpa)	18.6	17.9	18.0	18.9	
Adhesive property (index)	84	. 98	85	94	- 1
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[0053] The kinds of the processing acids in the above table are shown below:

- An ester compound of dehydrated castor oil fatty acid having a conjugated linoleic acid content of 45 % A: with 1,4-butylene glycol
 - An ester compound of dehydrated castor oil fatty acid having a conjugated linoleic acid content of 45 % B: with 2-ethylhexanol
 - An ester compound of dehydrated castor oil fatty acid having a conjugated linoleic acid content of 45 % with benzyl alcohol
 - An ester compound of dehydrated castor oil fatty acid having a conjugated linoleic acid content of 45 % D:
- with methyl alcohol An ester compound of dehydrated castor oil fatty acid having a conjugated linoleic acid content of 45 % E:
- with diethylene glycol A mixture of ester compounds of dehydrated castor oil fatty acid having a conjugated linoleic acid content F٠
- of 45 % with trimethylolpropane or 1,4-butylene glycol (30 % by weight/70 % by weight) An ester compound of dehydrated castor oil fatty acid having a conjugated linoleic acid content of 50 % G:
- with 1,4-butylene glycol
- H: An ester compound of oleic acid with 1,4-butylene glycol
- 1: An ester compound of oleic acid with 2-ethylhexanol

J: An ester compound of oleic acid with benzyl alcohol

An ester compound of dehydrated castor oil fatty acid having a conjugated linoleic acid content of 35 %

with 1,4-butylene glycol

K:

Adhesive: Manobond C22.5 manufactured by Rhone Poulenc Co., Ltd. 6PPD; N-(1,3-dimethylbutyl)-N'-phenyl-p-phenylenediamine

DCBS: N,N-dicyclohexyl-2-benzothiazolyl sulfenamide

Comments on Table 1 and Table 2

As apparent from the results shown in Table 1 and Table 2 described above, it has been found that the rub-[0054] ber compositions in Examples 1 to 9 falling in the scope of the present invention is markedly excellent in the processability in the unvulcanized state and have the excellent low heat-generating property without deteriorating the mechanical property (tensile strength) after vulcanization as compared with those in Comparative Examples 1 to 6 falling outside the scope of the present invention, and that in Examples 10 to 18, the adhesion between steel and vulcanized rubber is excellent as compared with those in Comparative Examples 7 to 12 falling outside the scope of the present invention. To specifically observe the examples, Examples 1 to 9 are cases where the ester compounds obtained by a reaction of various fatty acids having a conjugated linoleic acid content of 40 % by weight or more with various alcohols are blended in the blending amounts falling in the scope of the present invention respectively with 100 parts by weight of the rubber component comprising natural rubber and polybutadiene rubber. Particularly in Example 9, the blending amount of the ester compound is set to 5 parts by weight, and this is a case where the amount is increased by 2.5 times based on that of Example 1 and by 10 times based on that of Example 8. It can be found that also in this case, the processability is further improved without deteriorating the mechanical property and the low heat-generating property after vulcanization. It has been confirmed that the rubber compositions prepared in Examples 1 to 9 can suitably be used for tread rubber of a tire.

§ 1056] On the other hand, Examples 10 to 18 are cases where the ester compounds obtained by a reaction of various tatty acids having a conjugated linolist caid content of 40 % by weight or more with various alcohols are blended in the blending amounts falling in the scope of the present invention respectively with 100 parts by weight of the rubber component comprising natural rubber and where an adhesive (Manobond C22.5 manufactured by Pflone Poulenc Co., Ltd.) is further blended. Particularly in Example 18, the blending amount of the ester compound is set to 5 parts by weight, and this is a case where the amount is increased by 2.5 times based on that of Example 17. It can be found that also in this case, the processibility is further improved without deteriorating the mechanical property and the low heat-generating property after vulcanization and the adhesion between steel and vulcanized rubber.

posert, the compositions are blended based on 100 parts by weight of the nubber component comprising natural rubber and polybutadene rubber in Comparative Examples 1 to 6 and based on 100 parts by weight of the rubber component comprising natural rubber and polybutadene rubber in Comparative Examples 1 to 6 and based on 100 parts by weight of the rubber component comprising natural rubber in Comparative Examples 7 to 12. Comparative Examples 1 and 7 are cases (controls) where the seter compound according to the present invention is not blended; Comparative Examples 6 and Comparative Examples 8 to 10 are cases where the ester compounds obtained by reacting oleic acid with various alcohols are blended; Comparative Examples 5 and 11 are cases where the ester compounds obtained by reacting a fatty acid (dehydrated castor oil fatty acid) having a conjugated tinoleic acid content of 33 % by weight, which is less than 40 % by weight, various alcohols are blended; and Comparative Examples 6 and 12 are cases where the ester compounds obtained by reacting a fatty acid having a conjugated linoleic acid content of 40 % by weight on with various alcohols are blended but the blending amount thereof is 0.1 part by weight, which is less than 0.2 part by weight. It are a be found that in any of the cases of these Comparative Examples 1 and 12, the effects of the present invention can not be exerted.

Claims

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- 60 1. A rubber composition characterized by containing, per 100 parts by weight of a vulcanizable rubber component, 0.2 to 20 parts by weight of an ester compound obtained by a reaction of alcohols with a fatty acid having a conjugated lirolete acid content of 40% by weight or more.
 - The rubber composition as described in claim 1, wherein monohydric alcohol and/or dihydric alcohol account for 60 % by weight or more of the alcohols.
 - The rubber composition as described in claim 1 or 2, wherein the ester compound has a viscosity of 140 mPa ·s
 or less at 25°C.

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